

### **Mechanical Engineering Department**

Academic Year – 2020-21	Class: SE
Semester – I	Date : 17/08/2020
CO: CO1	PO: PO1, PO10

#### **Innovative Teaching Methods**

**Title of Innovation method/activity:** Innovative Teaching Learning Method (Preparation and presentation of Models)

Link shared to the students:

- 1. Name of Faculty: Mr. N.S. Gaikwad
- 2. Subject: Material Science
- 3. Objective of Method:
  - I. Prepared construction, working and application of crystal models
  - II. Describe the types of crystal structure
  - III. Identify no. of atoms per unit cell, coordination number.
  - IV. Determine atomic packing factor.
  - V. Develop practical and presentation skill

### 4. Topic Covered through Activity:

Understand construction and lattice parameters of crystal structures.

### 5. Description of method with Benefits (8 – 10 lines):

Different types of Crystal structures will be prepared by students by using available material and to give a presentation. The hands-on activity is designed for students to prepared different crystal structures.

Benefits of method:

- It helps students to better understanding crystal structure and how its parameters
- Students have to understand basic concepts of crystal structure and find Navg, Ligancy no .and APF.
- It helps to students to share his ideas with classmates and builds oral communication skills.

## The method:

Monitor and support students as they work through the following in this method:

1. Ask students to make a model of any crystal structure by using available material

2. Prepared model is developed by individually.

3. All groups are asked to give presentation of prepared model.

4. Teacher examined the presentation of each group and asks questions related to type and Navg.,co-ordination number, APF and various parameters of crystal structure.

Roles and Responsibilities

- Teacher
  - Suggest available material for development of crystal model.
  - Provide the study material of different crystal structures and appropriate guide lines at every stage of making models.
  - Remain available during the completion of task.
  - Prepare assessment methodology.
- Student
  - Go through all the material provided on model.
  - Once model is selected, understand it and discuss individually
  - Actively participate in presentation and contribute by means of discussion

# 6. Assessment Tools

Sr. No.	Rubrics	Marks
1	Model preparation	4 M
2	Understanding	3 M
3	Presentation skill	3 M

# 7. Evaluation sheet of attendee

Sr.	Roll	Name of	Model	Understanding	Presentation	Score
No.	No.	students	preparation(4)	(3)	skill (3)	out of 10
1		Harsh Dattatray	3	2	2	7
	1	Aher				
2		Ganesh Umesh	3	2	2	7
	2	Ahirrao				
3		Sidhhant Sunil	3	2	2	7
	5	Bachhav				/
4		Tejas Santosh	3	2	2	7
	7	Baviskar				1
5		Sushant Rajendra	3	2	2	7
	10	Bodke				
6		a 11	4	3	3	10
	1.1	Swapnil				
	11	Tukaram Borade			2	10
1	1.0	Samarth	4	3	3	10
	12	Bhagwat Boraste				
8		Vishal Ashok	3	3	3	9
-	13	Budhawant				
9		Rushikesh Manoj	3	3	2	8
	15	Chavan	-	-	-	
10		Ankush Santosh	3	3	2	8
	16	Derle				
11		Rohit Ramesh	3	3	2	8
	19	Dugaje				
12		Sakshi Vishwas	4	2	2	8
	20	Gadakh				
13		Vishwajeet	3	3	3	9
	21	Rajendra Gadakh				
14		Onkar Hemant	3	3	3	9
	22	Gaikwad				
15		Prasad Dipak	3	3	3	9
	25	Hire				
16		Prathamesh	4	3	3	10
	26	Dinesh Jadhav				
17		Umesh Shriram	3	3	3	9
	27	Jadhav				

18		Amey	4	3	3	10
		Appasaheb				
	28	Kadam				
19		Rutuj Sanjay	4	3	2	9
	32	Khairnar				
20		Yashodeep Arun	3	3	3	9
	33	Khairnar				
21		Vaibhav Ashok	4	3	3	10
	36	Kirtiwar				
22		Gaurav	4	3	2	
		Murlidhar				8
	37	Kothawade				
23	38	Pratik Ghotekar	2	3	2	7
24		Abhishek Sharad	4	3	2	9
	39	Kotkar				
25		Vaibhav	4	2	2	8
		Dnyaneshwar				
	40	Kushare				
26		Yuvaraj	4	3	2	9
		Ramchandra				
	41	Labhade				
27		Aniket Kailas	4	3	2	9
	42	Lahane				
28		Aniket Subhash	4	3	2	9
	43	Lawand				
29		Jay Rajendra	4	3	3	10
	44	Mahale				
30		Omkar Manoj	4	3	3	10
	45	Mali				
31		Chaudhari Uday	4	3	3	10
	48	Yuvraj				10
32		Gunjal Shivraj	3	3	3	9
	51	Gajendra				,
33		Jagtap Himanshu	4	3	3	10
	53	Shekhar				
34		Mondhe	2	3	2	7
	56	Purushottam				
35	57	Patale Rohit	2	2	2	6
36		Shubham	4	2	2	8
	59	Panditrao Jadhav				
37		Piyush	4	3	2	9
	60	Sonawane				

Student responses: Uploaded as a separate excel sheet

# 8. Activity Picture



9. For review and critique contact: e-mail address of faculty and HOD gaikwad.nilesh@kbtcoe.org, hod.mech@kbtcoe.org

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